

SYNTHESIS OF BENZO[a]FURO[2,3-g]QUINOLINZINES FROM  
2-METHYL-2-CINNAMOYLOXIRANE

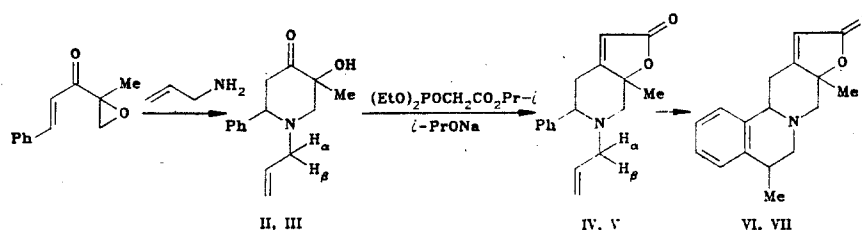
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Stereoisomeric 1e-allyl-3-hydroxy-3-methyl-6e-phenyl-4-piperidones were obtained by reaction of 2-methyl-2-cinnamoyloxirane with allylamine. Their PO-olefinization and subsequent heating in 96% H<sub>2</sub>SO<sub>4</sub> leads to 3a,4,6,7,11b,12-hexahydro-3a,7-dimethyl-2-oxobenzo[a]furo[2,3-g]quinolizines.

Interest in the chemistry of benzo[a]furoquinolizines is due to the similarity of their structure to that of natural and related active compounds [1-3]. The only method until recently for constructing the benzo[a]furo[2,3-g]quinolizine skeleton is by reductive photocyclization of enamides of the isoquinoline series [3].

We have proposed a new path of synthesis of compounds of this type [4], based on the readily available cinnamoyloxirane I. Thus, in the reaction of compound I with allylamine, stereoisomeric piperidones II, III were obtained, which were converted according to the Wittig-Horner reaction into lactones IV, V. On heating in 96% H<sub>2</sub>SO<sub>4</sub> the latter cyclized into quinolizidines VI, VII.



II 3e-OH, 3a-Me, 6e-Ph; III 3a-OH, 3e-Me, 6e-Ph; IV 5e-Ph; V 5a-Ph; VI 11b(α)-H, 3a(β) and 7(α)-Me; VII 11b(β)-H, 3a(β) and 7(β)-Me

The structure of all the synthesized compounds was confirmed by the data of elemental analysis, IR, PMR, NMR and mass spectra (Tables 1 and 2). The structure of 1e-allyl-3-hydroxy-4-piperidones II, III conforms well with the literature data [5, 6]. The frequency values of the absorption maxima in the IR spectra, corresponding to the C=C and C=O bonds of lactones IV-VII, are characteristic of unsaturated γ-lactones. The chair-conformation of the six-membered rings in compounds II-VII and the relative orientation of their substituents follow from the spin-spin coupling constants in the PMR spectra (Table 1) [8]. The types of coupling of rings B and C in quinolizidines VI, VII is also confirmed by the presence or absence of Bohlmann bands in the IR spectra in the 2700...2850 cm<sup>-1</sup> region [9]. The proposed conformation of compound VI with an α-orientation of the 11b-H atom conforms with the presence of these absorption bands, and also with the value of the chemical shift of the proton under consideration in the PMR spectrum [10]. In the IR spectrum of quinolizidine VII, the Bohlmann bands are absent, and in the case of a chair-conformation of the piperidine ring, the cis-B/C coupling appears to be the only one possible. The shift to the strong field of most of the carbon atoms signals of the quinolizidine fragment in the <sup>13</sup>C NMR spectra (Table 2) during the cis-B/C coupling, which is noted for benzo[a]quinolizidines VI, VII, is significant [11].

To investigate the pharmacological activity, we studied the acute toxicity, the neurotropic and analgesic properties of methanesulfonate VI and hydrochloride VII in tests on white mice.

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TABLE 1. Characteristics of Compounds II-VII

Com- pound	Empirical formula	mp, °C	IR spectrum, cm <sup>-1</sup>	PMR spectrum, $\delta$ , ppm (SSCC, Hz)	M <sup>a</sup>	Yield, %
II	C <sub>15</sub> H <sub>19</sub> NO <sub>2</sub>	45...46	1645, 1720, 3505	1.63 (s, 3-CH <sub>3</sub> ); 2.24 (d, 2-H <sub>a</sub> ; 11.3); 2.52 (d, d, 5-H <sub>a</sub> ; 13.5; 3.5); 2.55 (d, d, H <sub>a</sub> ; 14.0; 7.0); 2.83 (d, d, 5-H <sub>a</sub> ; 13.5; 11.5); 3.16 (d, d, t, H <sub>b</sub> , 14.0; 4.5; 2.0); 3.27 (d, 2-H <sub>a</sub> ; 11.3); 3.47 (d, d, 6-H <sub>a</sub> ; 11.5; 3.5); 3.83 (s, OH); 5.05...5.16 (m, C=CH <sub>2</sub> ); 5.66...5.86 (m, CH=C); 7.23...7.40 (m, 5H arom.)	245	81
III	C <sub>15</sub> H <sub>19</sub> NO <sub>2</sub>	Oil	1645, 1715 sh., 1730, 3495	1.28 (s, 3-CH <sub>3</sub> ); 2.37 (d, 2-H <sub>a</sub> ; 12.0); 2.48 (d, d, 5-H <sub>a</sub> ; 14.5; 4.0); 2.75 (d, d, H <sub>a</sub> ; 14.5; 7.5); 3.04 (d, d, 5-H <sub>a</sub> ; 14.5; 10.5); 3.09 (d, 2-H <sub>a</sub> ; 12.0); 3.12 (d, d, t, H <sub>b</sub> ; 14.5; 4.5; 2.0); 3.59 (d, d, 6-H <sub>a</sub> ; 10.5; 4.0); 3.40...3.72 (OH); 5.02...5.13 (m, C=CH <sub>2</sub> ); 5.65...5.84 (m, CH=C); 7.17...7.37 (m, 5H arom.)	245	80
IV	C <sub>17</sub> H <sub>19</sub> NO <sub>2</sub>	80...81	1655, 1765, 1775 sh.	1.68 (s, 7a-CH <sub>3</sub> ); 2.07 (d, 7-H <sub>a</sub> ; 10.8); 2.57 (d, d, d, 4-H <sub>a</sub> ; 13.5; 10.8; 1.8); 2.59 (d, d, H <sub>a</sub> ; 14.0; 7.5); 2.84 (d, d, 4-H <sub>a</sub> ; 13.5; 3.5); 3.13 (d, d, t, H <sub>b</sub> ; 14.0; 4.5; 2.0); 3.20 (d, d, 5-H <sub>a</sub> ; 10.8; 3.5); 3.43 (d, 7-H <sub>a</sub> ; 10.8); 5.03...5.12 (m, C=CH <sub>2</sub> ); 5.62...5.85 (m, CH=C); 5.68 (d, 3-H; 1.8); 7.26...7.40 m. d. (m, 5H arom.)	269	65
V	C <sub>17</sub> H <sub>19</sub> NO <sub>2</sub>	67...67.5	1650, 1770	1.68 (s, 7a-CH <sub>3</sub> ); 2.71 (d, 7-H <sub>a</sub> ; 11.0); 2.84 (d, d, H <sub>a</sub> ; 14.2; 7.0); 2.98 (d, 7-H <sub>a</sub> ; 11.0); 3.02 (d, d, 4-H <sub>a</sub> ; 14.0; 1.5); 3.09 (d, d, d, 4-H <sub>a</sub> ; 14.0; 7.1; 1.8); 3.26 (d, d, t, H <sub>b</sub> ; 14.5; 5.8; 1.6); 4.29 (d, d, 5-H <sub>a</sub> ; 7.1; 1.5); 5.10...5.20 (m, C=CH <sub>2</sub> ); 5.72...5.82 (m, CH=C); 5.81 (d, 3-H; 1.8); 7.12...7.19 (m, 2H arom.); 7.24...7.38 (m, 3H arom.)	269	62
VI	C <sub>17</sub> H <sub>19</sub> NO <sub>2</sub>	106...107	1655, 1768, 1780 sh., 2765, 2805	1.27 (d, 7-CH <sub>3</sub> ; 7.0); 1.61 (s, 3a-CH <sub>3</sub> ); 2.33 (d, 4-H <sub>a</sub> ; 10.8); 2.35 (t, 6-H <sub>a</sub> ; 11.5); 2.51 (d, d, d, 12-H <sub>a</sub> ; 12.5; 10.5; 1.8); 2.98 (d, d, 6-H <sub>a</sub> ; 11.5; 5.2); 3.20...3.28 (m, 7-H <sub>a</sub> ); 3.29 (d, 4-H <sub>a</sub> ; 10.8); 3.33 (d, d, 11b-H <sub>a</sub> ; 10.5; 3.0); 3.40 (d, d, 12-H <sub>a</sub> ; 12.5; 3.0); 5.81 (d, 1-H; 1.8); 7.16...7.41 (m, 4H arom.)	269	63
VII	C <sub>17</sub> H <sub>19</sub> NO <sub>2</sub>	129...130	1658, 1770, 1780 sh.	1.20 (d, 7-CH <sub>3</sub> ; 6.3); 1.65 (s, 3a-CH <sub>3</sub> ); 2.57 (d, 4-H <sub>a</sub> ; 10.5); 2.96 (t, 6-H <sub>a</sub> ; 11.5); 2.97 (d, 4-H <sub>a</sub> ; 10.5); 2.98 (d, d, d, 12-H <sub>a</sub> ; 13.5; 5.6; 1.8); 2.96...3.04 (m, 7-H <sub>a</sub> ); 3.17 (d, d, 6-H <sub>a</sub> ; 11.5; 5.2); 3.46 (d, d, 12-H <sub>a</sub> ; 13.5; 1.6); 4.49 (d, d, 11b-H <sub>a</sub> ; 5.6; 1.6); 5.61 (d, 1-H; 1.8); 7.08...7.27 (m, 4H arom.)	269	70

TABLE 2. Chemical Shifts in <sup>13</sup>C NMR Spectra of Compounds VI and VII

Com- pound	$\delta$ , ppm											
	C <sub>(1)</sub>	C <sub>(2)</sub>	C <sub>(3a)</sub>	C <sub>(4)</sub>	C <sub>(6)</sub>	C <sub>(7)</sub>	C <sub>(7a)-C<sub>(11a)</sub></sub>	C <sub>(11b)</sub>	C <sub>(12)</sub>	C <sub>(12a)</sub>	3a-CH <sub>3</sub>	7-CH <sub>3</sub>
VI	112.62	172.10	84.18	66.69	59.61	32.99	124.68; 126.03; 126.64; 126.87; 135.91; 139.68	64.29	34.07	173.41	22.20	17.82
VII	114.12	172.03	84.27	58.80	58.32	26.56	124.88; 126.12; 127.07; 127.69; 132.52; 139.57	58.58	28.64	170.54	22.18	19.31

The acute toxicity index ( $LD_{50}$ ) on intravenous administration of the above salts was equal to 181.4 and 114.6 mg/kg, respectively, i.e., in accordance with the accepted classification these compounds can be classed as slightly toxic.

When administered to the mice in doses comprising 20% of  $LD_{50}$  these compounds manifest a distinct general neurotropic action: they suppress the orienting reaction; increase the survival time of test animals after the administration of strychnine and corazole; intensify the reserpine hypothermia; prolong the analgesic effect of morphine, although by themselves they do not alter the threshold of pain sensitivity to a temperature dependent irritant.

Thus, a preparative method has been developed for the synthesis of benzo[a]furo[2,3-g]quinolizines, which are of definite interest for the synthesis of new biologically active compounds.

#### EXPERIMENTAL

The IR spectra of solutions of the compounds in  $C_2Cl_4$  were run on a UR-20 spectrophotometer. The PMR and  $^{13}C$  NMR spectra were obtained in  $CDCl_3$  on a Bruker WM-360 spectrometer, using HMDS as internal standard. The mass spectra were recorded on a Varian MAT-311 mass-spectrometer at 70 eV. The characteristics of compounds II-VII are given in Table 1. The data of the elemental analysis of the compounds obtained agree with the calculated values.

1e-Allyl-3a-hydroxy-3a-methyl-6e-phenyl-4-piperidone (II). A 50.0 g portion (0.27 mole) of cinnamoyloxirane I was dissolved in 300 ml of dioxane, and 50 ml of water and 30 g (0.52 mole) of allylamine were successively added. The reaction mixture was allowed to stand for 24 h at 18...20°C, and then was evaporated. The residue was dissolved in 200 ml of 10% HCl, and the solution was allowed to stand overnight. The aqueous solution was then filtered and the filtrate was made alkaline with  $NaHCO_3$ . The precipitate that separated out was extracted with ether, the ether solution was dried over  $Na_2SO_4$ , evaporated, and the residue was crystallized from pentane.

1e-Allyl-3a-hydroxy-3e-methyl-6e-phenyl-4-piperidone (III). An 8 g portion (0.21 mole) of allylamine was added at 150...160°C to a solution of 18 g (0.096 mole) of cinnamoyloxirane I in 50 ml of benzene. The reaction mixture was allowed to stand at the above temperature for 2 h and was then treated as described in the preceding experiment.

6e-Allyl-8a-methyl-2-oxo-5-phenylfuro[2,3-c]piperidine (IV,V). A 28.6 g portion (0.12 mole) of isopropyl diethylphosphonoacetate was added to a solution of sodium isopropylate, prepared by dissolution of 2.53 g (0.11 mole) of sodium in 250 ml of isopropanol, and then while cooling the mixture with tap water, 21.9 g (0.1 mole) of piperidone II, III was added. The mixture was stirred to the complete dissolution of piperidone; it was then acidified with acetic acid, the solvent was evaporated, and the residue was dissolved in ether and washed with a saturated solution of  $NaHCO_3$  and water. After evaporation of the solvent, the residue was dissolved in 100 ml of isopropanol, 16.4 g (0.20 mole) of sodium isopropylate was added, and the mixture was allowed to stand for 5 h at 20...25°C. The mixture was then acidified with acetic acid, the solvent was evaporated, and the residue was dissolved in ether. The ether solution was washed with a saturated solution of  $NaHCO_3$  and water, dried over  $Na_2SO_4$ , evaporated and the residue was crystallized from isopropanol.

3a,4,6,7,11b,12-Hexahydro-3a,7-dimethyl-2-oxobenzo[a]furo[2,3-g]quinolizines (VI, VII). A mixture of 1 g (3.8 mmole) of lactone IV, V and 2 ml of 96%  $H_2SO_4$  was allowed to stand for 2 h 30 min at 110...115°C. It was then diluted with water (30 ml), and allowed to stand for another hour at the same temperature. The solution was cooled, made alkaline with  $Na_2CO_3$ , extracted with ether, and the extract was dried over  $Na_2SO_4$ . The solvent was evaporated, and the residue was crystallized from isopropanol.

3a,4,6,7,11b( $\alpha$ ),12-Hexahydro-3a( $\beta$ ),7( $\alpha$ )-dimethyl-2-oxobenzo[a]-furo[2,3-g]quinolizine methanesulfonate ( $IV \cdot CH_3SO_3H$ ,  $C_{17}H_{19}NO_2 \cdot CH_4O_3S$ ). A 1.06 g portion (11 mmoles) of methanesulfonic acid was added to a solution of 2.69 g (10 mmoles) of lactone IV in 10 ml of acetone. The precipitated salt was filtered and washed with cold acetone. Yield, 3.2 g (88%) of methanesulfonate, mp 193...194°C.

3a,4,6,7,11b( $\beta$ ),12-Hexahydro-3a( $\beta$ ),7( $\beta$ )-dimethyl-2-oxobenzo[a]-furo[2,3-g]quinolizine hydrochloride ( $V \cdot HCl$ ,  $C_{17}H_{19}NO_2 \cdot HCl$ ). Dry HCl was passed through a solution of 2.69 g (10 mmoles) of lactone V in 20 ml of ether to pH 4...5. The precipitated hydrochloride was re-crystallized from isopropanol. Yield 2.8 g (91%) of the salt, mp 208-210°C.

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